

## Research Article

# Radiosynthesis of [ $^{123}\text{I}$ ]N-(3-iodoprop-(2*E*)-enyl)-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropine as a potential SPET tracer for dopamine transporter

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## Summary

The radiosynthesis of a novel tropane derivative [ $^{123}\text{I}$ ]KUC-25019, [[ $^{123}\text{I}$ ]N-(3-iodoprop-(2*E*)-enyl)-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl)nortropine], a potential inhibitor of the dopamine transporter is reported. The synthetic routes include the preparation of standard reference, the stannyl precursor and the  $^{123}\text{I}$ -labeling synthesis. The no-carrier-added  $^{123}\text{I}$ -labeling has about 20% yield, the specific activity of [ $^{123}\text{I}$ ]KUC-25019 is > 107 GBq/ $\mu\text{mol}$  and the radiochemical purity of [ $^{123}\text{I}$ ] KUC-25019 is > 95%. Copyright © 2002 John Wiley & Sons, Ltd.

**Key Words:** KUC-25019; dopamine transporter; tropane analog; SPET:  $^{123}\text{I}$

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Contract/grant sponsor: National Technology Agency (TEKES, Helsinki).

Contract/grant sponsor: Kuopio University Hospital; contract/grant numbers: 5031302, 5031301.

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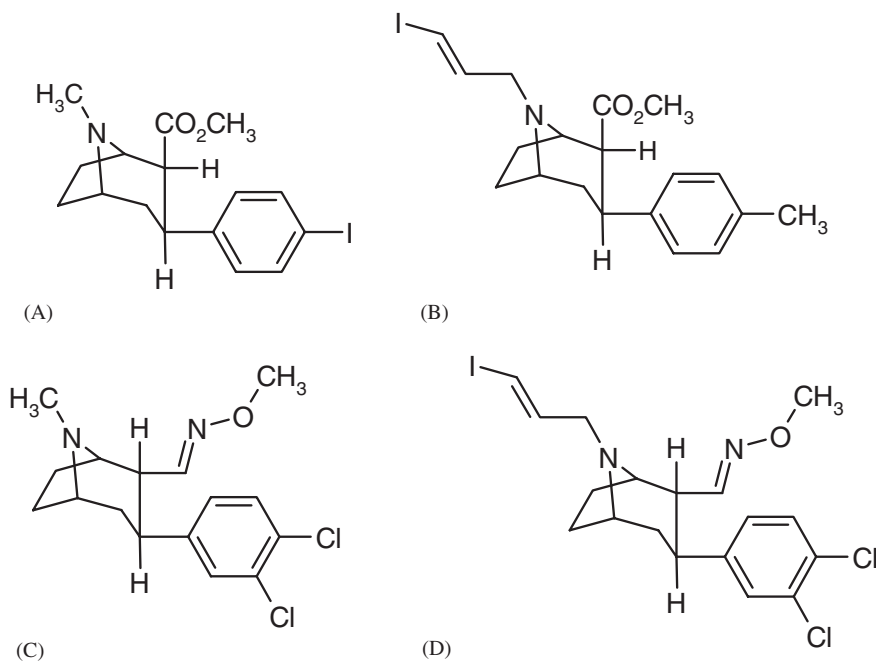
Received 7 January 2002

Revised 30 April 2002

Accepted 31 May 2002

## Introduction

Recently, the imaging of the dopamine transporter (DAT) *in vivo* with single-photon emission tomography (SPET) or positron emission tomography (PET) has increased greatly because of a successful development of a numerous of new tropane analogs.<sup>1</sup> Among them, [<sup>11</sup>C]β-CIT (WIN 35-428)<sup>2</sup> and [<sup>123</sup>I] β-CIT<sup>3</sup> (compound A, in Figure 1) have been used in clinical brain studies, such as in Parkinson's disease. Recently it is found that a new compound [<sup>123</sup>I]PE2I<sup>4</sup> (compound B, in Figure 1) is a selective ligand for imaging the DAT with different kinetic properties than [<sup>123</sup>I]β-CIT. However, cocaine analogs containing 3-carboxylic acid functionalities in the exo-(β-) conformation are known to exhibit powerful stimulant effects.<sup>5</sup> A new cocaine congeners NS 2214 (compound C in Figure 1) having 3-substituted aldoxime prosthetic group in the endo-(α-) configuration has been reported recently to have high DAT affinity and a low toxicity.<sup>5,6</sup>

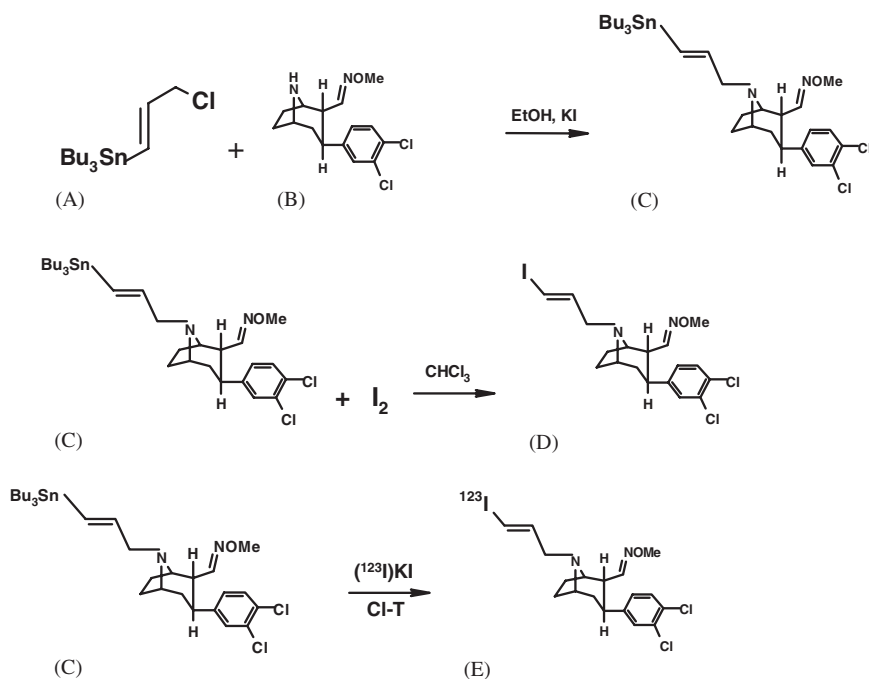


**Figure 1.** Chemical structures of β-CIT (A), PE2I (B), NS 2214 (C) and KUC-25019 (D)

We have synthesized a new NS2214 analog, *N*-(3-iodoprop-(2*E*)-enyl)-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropane (KUC-25019, compound D in Figure 1). This ligand with a 3-substituted aldoxime prosthetic group in the endo-( $\alpha$ -) configuration may have lower toxicity than those cocaine analogs (such as  $\beta$ -CIT,  $\beta$ -CFT) containing 3-carboxylic acid functionalities in the exo-( $\beta$ -) conformation. KUC-25019 can be labeled with  $^{123}\text{I}$  and may be useful in the imaging of DAT.

## Materials and Methods

All chemicals and solvents were of commercial quality and were purified following the standard procedures if necessary. The alkylating agent 3-(tributylstannyl)prop-(2*E*)-enyl chloride (compound A in Figure 2) was prepared by chlorination of pure 3-(tributylstannyl)prop-(2*E*)-en-1-ol with triphenylphosphine and carbon tetrachloride.<sup>7</sup> The pure



**Figure 2.** Synthesis of the precursor (C), KUC-25019 (D) and the radiolabelled KUC-25019 (E)

(*E*)-stannyl alcohol was obtained by hydrostannylation of propargyl alcohol<sup>8</sup> followed by flash chromatography purification using petroleum ether/ethyl acetate (9/1).<sup>9</sup> 2 $\alpha$ -(Imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropane was kindly offered by Neurosearch A/S (Smedeland, Denmark) and was confirmed by NMR.

<sup>1</sup>H NMR spectra were obtained on a Bruker 400 MHz spectrometer. CDCl<sub>3</sub> was used as a solvent, and the chemical shifts were reported in ppm with a reference to internal TMS. Flash chromatography was used for routine purification of reaction products using silica gel (230–400 mesh).

*N*-[3-(Tri-*N*-butylstannyl)prop-(2*E*)-enyl]-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropane

2 $\alpha$ -(Imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropane (compound B in Figure 2) 32 mg (0.10 mM) and 3-(tributylstannyl)prop-(2*E*)-enyl chloride (compound A in Figure 2, 55 mg, 0.15 mM) were dissolved into 10 ml absolute EtOH containing 1.0 ml dry triethyl amine and a catalytic amount of KI. The mixture was refluxed under nitrogen atmosphere for 14 h. The solvent was evaporated and the residue was purified by flash chromatography (petroleum ether 40–65°C:AcOEt, 8:2) to give a colorless oil product (22.5 mg, 35%).

*N*-[3-Iodoprop-(2*E*)-enyl]-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichloro-phenyl) nortropane

Stannyl derivative (21 mg) (compound C in Figure 2, 0.033 mM) was dissolved in 5 ml dry CHCl<sub>3</sub>, and the resulting mixture was cooled in ice-water bath. A solution of iodine in CHCl<sub>3</sub> (0.5 M) was then added dropwise to the stirred mixture until a color solution resulted. The reaction solution was washed with saturated NaCl solution, dried (anhydrous Na<sub>2</sub>SO<sub>4</sub>) and evaporated. The residue was purified by flash chromatography (Et<sub>2</sub>O) to give an oil product (7.5 mg, 47%).

*No-carrier-added (NCA) [<sup>123</sup>I] N*-(3-iodoprop-(2*E*)-enyl)-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropane

NCA <sup>123</sup>I solution (130  $\mu$ l) (15.2 mCi, specific activity 35–280 Ci/mg, PSI, Switzerland) was added to the stannyl precursor solution (100  $\mu$ g in 100  $\mu$ l EtOH) in a reaction vial. After addition of 100  $\mu$ l 0.2 M HCl,

100  $\mu$ l chloramine-T (1 mg/ml) was added. The reaction mixture was kept at room temperature. After 5 min, 300  $\mu$ l of HPLC solvent (acetonitrile/10 mM phosphoric acid, 1/1 v/v) was added and the reaction mixture was injected into the semipreparative HPLC RP18 (HPLC Technology, Macclesfield, UK) column (25 cm  $\times$  4.6 cm) using above HPLC solvent system at 4 ml/min. The radioactive peak with a retention time similar to a reference standard was collected. After evaporation, the residue was dissolved in ethanol and phosphate buffer in order to formulate an injection solution. The product was sterilized by filtration through a 0.2- $\mu$ m filter. The final product contained < 15 % (w/w) ethanol.

#### Radiochemical Purity Determination

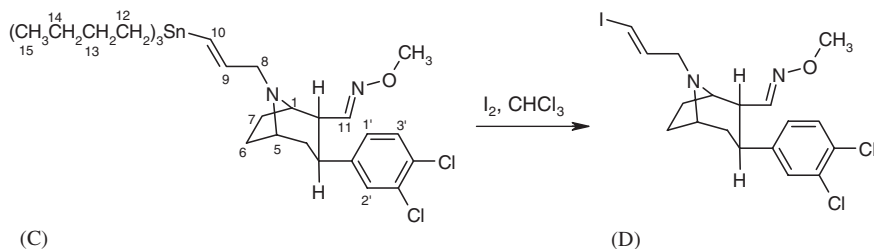
The radiochemical purity of the tracer was assayed with a HPLC system consisting of a Packard (Packard Instrument Co., Meriden, CT, USA) radioisotope detector with computer data acquisition.<sup>10</sup> The HPLC system is the same as above.

## Results and Discussion

### *Organic synthesis and radiolabeling*

The tributyltin precursor (compound C in Figure 2) was prepared by reacting 3-(tri-*n*-butylstannyl)prop-(2*E*)-enyl chloride (compound A in Figure 2) with nortropine (compound B in Figure 2) by the same method as reported.<sup>11</sup> The alkylating agent A was prepared by chlorination of pure 3-(tri-*n*-butylstannyl)prop-(2*E*)-en-1-ol with triphenylphosphine and carbon tetrachloride.<sup>7,9</sup> The pure (*E*)-stannyl alcohol was obtained by hydrostannylation of propargyl alcohol<sup>8</sup> followed by flash chromatography separation using petroleum ether/ethyl acetate (9/1).<sup>9</sup> The reference compound *N*-(3-iodoprop-(2*E*)-enyl)-2 $\alpha$ -(imino-methyl)-3 $\beta$ -(3',4'-dichlorophenyl) nortropine was obtained by iododestannylation of its stannyl precursor (compound C in Figure 2) by treatment with iodine in dry chloroform. All these organic synthesis reactions were well done, and Figure 3 showed the NMR data of the tributyl precursor and the standard KUC-25019.

The radiochemical labeling yield of [<sup>123</sup>I]KUC-25019 was about 20% calculated by dividing the radioactivity in the final product by the radioactivity in the NCA <sup>123</sup>I vial, corrected for decay ( $T_{1/2}$  13.2 h). The specific radioactivity of the product was > 107 GBq/ $\mu$ mol. The



C:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  (ppm) 0.90 (t,  $9\text{H}^{15}$ ,  $J = 7.3$  Hz), 1.26–2.10 (m, 24  $\text{H}^{12-14,4,6,7}$ ), 2.81–2.85 (m,  $2\text{H}^{2,3}$ ), 3.16 (d,  $2\text{H}^8$ ,  $J = 4.7$  Hz), 3.32 (m,  $1\text{H}^5$ ), 3.35 (dm  $1\text{H}^1$ ,  $J = 6.0$  Hz), 3.70 (s, 3H), 5.95–6.35 (m,  $2\text{H}^{9,10}$ ), 7.02 (d,  $1\text{H}^{11}$ ,  $J = 5.3$  Hz), 7.07 (dd,  $1\text{H}^1$ ,  $J = 8.3$  Hz,  $J = 2.1$  Hz), 7.30–7.35 (m,  $2\text{H}^{2,3}$ ).

D:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  (ppm) 1.40–2.05 (m, 6  $\text{H}^{4,6,7}$ ), 2.78–2.84 (m,  $2\text{H}^{2,3}$ ), 3.03 (d,  $2\text{H}^8$ ,  $J = 6.2$  Hz), 3.29 (m,  $1\text{H}^5$ ), 3.32 (dm,  $1\text{H}^1$ ,  $J = 6.1$  Hz), 3.72 (s, 3H), 6.28 (dt,  $1\text{H}^{10}$ ,  $J = 14.6$  Hz,  $J = 1.5$  Hz), 6.61 (ddd,  $1\text{H}^9$ ,  $J = 14.6$  Hz,  $J = 6.5$  Hz,  $J = 6.2$  Hz), 7.02 (d,  $1\text{H}^{11}$ ,  $J = 5.5$ ), 7.06 (dd,  $1\text{H}^1$ ,  $J = 8.4$  Hz,  $J = 2.0$  Hz), 7.31–7.36 (m,  $2\text{H}^{2,3}$ ).

**Figure 3.** NMR data for KUC-25019 and its stannyl precursor

radiochemical purity of [ $^{123}\text{I}$ ] KUC-25019 was >95% based on the radio-HPLC analysis of the purified product. Free  $^{123}\text{I}$  with a retention time of 1.6–1.9 min was identified as impurity. The product was eluted out with retention time of 10 min as determined by reference compound. The labeling yield (20%) was low probably due to the decomposition of the precursor (impurity was determined afterwards by NMR). The demonstrated radiosynthesis method was good enough to obtain sufficient amount of [ $^{123}\text{I}$ ]KUC-25019 for the animal study which will be done in the future.

### Acknowledgements

The study was supported by National technology Agency (TEKES, Helsinki) and Kuopio University Hospital (5031302, 5031301)

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